

Write the following Ionic Formulas

- a) Calcium Nitrate
- b) Aluminium Phosphate
- c) Sodium Nitrite
- d) Sodium Carbonate
- e) Potassium Sulfate
- f) Magnesium Phosphate
- g) Ammonium Hydroxide
- h) Ammonium Carbonate
- i) Barium Hydroxide
- j) Potassium Manganate (VII)
- k) Sodium di Chromate
- l) Iron (II) Hydroxide
- m) Copper (I) Phosphate
- n) Tin (II) Nitrate
- o) Aluminium Sulfide
- p) Calcium Nitride
- q) Sodium Oxide
- r) Aluminium Iodide
- s) Aluminium Hydrogen Carbonate
- t) Magnesium Hydrogen Sulfate
- u) Ammonium Sulfate
- v) Lithium Nitride